



Antioxidant Properties, α -Amylase and α -Glucosidase Inhibitory Activities of *Maesobotrya barteri* Leaves Extracts in Rats

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Keywords: Antioxidants, Alpha-glucosidase, Diabetes, Phytochemicals, *Maesobotrya barteri*.

Abstract: *Maesobotrya barteri* is widely used in Nigerian ethnomedicine to treat diabetes, arthritis, and infections. In this study, the methanol and aqueous leaf extracts' phytochemical constituents and antioxidant potentials were evaluated using standard procedures. At the same time, the enzyme inhibitory activity of methanol extract on α -amylase and α -glucosidase in rats was also investigated. The antioxidant properties of the extracts were evaluated by the 1,1-diphenyl-2-picrylhydrazyl (DPPH) radical scavenging activity and ferric reducing power (FRAP). Phytochemical screening of methanol and aqueous extracts revealed the presence of flavonoids, alkaloids, saponins, tannins, and cardiac glycosides. Total phenolics and flavonoids of the extracts ranged from 0.04 to 3.18 mg of GAE/g and from 27.70 to 57.70 mg of QE/g, respectively. Antioxidant analysis showed IC_{50} values of DPPH (192.95, 196.04, and 17.19 μ g/mL) and FRAP (30.48, 37.64, and 38.15 μ g/mL) for the methanolic extract, aqueous extract, and ascorbic acid, respectively. Assessment of the methanolic extract against α -amylase and α -glucosidase enzymes in rats at doses of 150, 300, and 450 mg/kg using starch, sucrose, and maltose as substrates, with acarbose as a reference drug, significantly reduced blood glucose levels ($p < 0.05$). These findings suggest that *M. barteri* leaf extract has antioxidant properties and inhibits both α -amylase and α -glucosidase enzymes in rats, likely due to the phytochemicals present in the extract.

Introduction

Diabetes and its related complications are increasingly recognized as significant health challenges linked to oxidative stress and inflammation in humans. The rising prevalence of insulin resistance and type 2 diabetes has become a global health crisis. In 2015, there were an estimated 415 million cases of diabetes (1, 2). By 2019, it was revealed that over 463 million people worldwide were affected by diabetes, with 19.4 million cases in Africa, representing approximately 4.7% of the adult population aged 20 to 79 years. Projections suggested this number could rise to 783 million by 2045 (1-4). To manage diabetes, various treatments, including synthetic drugs like metformin and phenformin, have been widely prescribed. While these medications effectively enhance peripheral glucose uptake and reduce hepatic glucose production,

long-term use can lead to undesirable side effects (5). In contrast, bioactive compounds in plants, fruits, and vegetables, which possess antidiabetic and antioxidant properties, have shown promise in alleviating diabetic symptoms and serve as valuable nutritional resources to combat oxidative stress.

Antioxidants are crucial in neutralizing free radicals and mitigating oxidative stress, which are linked to various diseases, including diabetes, cancer, cardiovascular issues, inflammation, thyroid disorders, and neurodegenerative conditions. Antioxidants help to prevent the formation of reactive oxygen species and capture harmful radicals while also repairing damaged nucleic acids, removing oxidized proteins, and restoring oxidized lipids with the aid of enzymes like hydrolases and phospholipases (6). Free radicals such

as reactive oxygen, nitrogen, and chlorine species are the major causes of health problems globally (7). These highly reactive entities, including hydroxyl radicals, nitric oxide radicals, hydrogen peroxide, superoxide anions, lipid peroxides, and various singlet oxygen molecules, can harm nucleic acids, proteins, enzymes, and other vital biomolecules, ultimately disrupting their structure and function (8).

An imbalance between prooxidants and antioxidants in the body leads to oxidative stress. Endogenous antioxidants act as free radical scavengers and lipid peroxidation inhibitors, such as superoxide dismutase, glutathione, uric acid, melatonin, metal-binding proteins, and polyamines. However, they require support from exogenous antioxidants to effectively maintain the body's functions (9, 10). Bioactive compounds, particularly those rich in polyphenols found in fruits, vegetables, and spices, are reliable sources of these antioxidants. Research has shown that diets high in vitamins, carotenoids, and phenolic compounds can protect human cells from prooxidant-related diseases (8, 11). To explore these benefits, various plant species have been screened for their antioxidant activities to expand natural antioxidants' ability to treat health conditions.

Maesobotrya barteri, a shrub from the Euphorbiaceae family, is native to several African regions, including the rainforest areas of Sierra Leone, Southern Nigeria, and Western Cameroon. It bears succulent white berries, commonly referred to as "squirrel cherry" in English and "Nyanyated" by the Ibibio people of Akwa Ibom State, Nigeria. However, some reports mention a black-purple variety. Ethnopharmacologically, the plant has been used to treat diabetes, malaria, dysentery, arthritis, mumps, and rheumatism (12, 13). Its twigs are used as chewing sticks, roots are infused in gin for arthritis treatment, and stems are employed for fencing and supporting yam tendrils (12, 13). A recent study has reported the presence of alkaloids, flavonoids, terpenoids, saponins, and tannins in the leaves of *M. barteri* (14).

Despite the fact that *M. barteri* is used traditionally in the southern part of Nigeria for the treatment of diabetes, malaria, dysentery, arthritis, mumps, and other ailments, there is scanty literature information on the antidiabetic activity of this plant. Inhibition of α -amylase and α -glucosidase study is one of the modes of antidiabetic activity. This study was therefore designed to evaluate the antidiabetic and antioxidant activities of *M. barteri* leaf extracts by analyzing their phytochemical components, DPPH, and FRAP, as well as evaluating the effects of the methanolic extract on alpha-amylase and alpha-glucosidase activities in rats. The findings of this research will contribute to the scientific understanding of the plant's antidiabetic potential and its phytochemical composition.

Methodology or Experimental Section

Materials

The materials used include *M. barteri* leaf extract, stirrer, beakers, 1 mL syringe, iodine-potassium iodide, aluminum chloride, sodium hydroxide, hydrochloric acid, ferric chloride, sodium carbonate, sulphuric acid, phosphate buffer, and potassium ferricyanide, oral gastric gavage, weighing balance, gloves, scissors, glucometer and strips (fine test), distilled water, acarbose, starch, sucrose, maltose (from Sigma Aldrich, USA).

Plant Materials

Plant Collection, Identification, and Preparation

M. barteri leaves were harvested from a forest in Ediene Attai Village in Oruk Anam Local Government Area of Akwa Ibom State, Nigeria, in March 2024. The plant identification was carried out at the Department of Botany and Ecological Study, Faculty of Biological Sciences, University of Uyo, Nigeria. The fresh leaves were washed with flowing water, air-dried at ambient temperature for two weeks, and reduced into a fine powder using a laboratory mill.

Plant Extraction

The method of Ouandaogo et al. (2023) was used to extract the plant sample (15). Ninety grams (90 g) of the finely powdered leaves were placed in conical flasks and extracted with 70% methanol. The flasks were placed on a flat plate mechanical shaker (model: platform ZD 881) and macerated for 14 hours at 25 °C. The resulting solution was filtered, and the filtrate was concentrated to dryness *in vacuo* to obtain the methanol extract. Another 90 g of the powdered leaves were macerated with distilled water at 65 °C according to the previous method to obtain the aqueous extract (13). The methanol and aqueous extracts were weighed, the percentage yield calculated as shown below, and the extracts were refrigerated until use.

Qualitative Phytochemical Screening

Preliminary Phytochemical Screening

Phytochemical test for flavonoids, alkaloids, saponins, tannins, cardiac glycosides, and anthraquinones were conducted according to standard methodologies (13).

Flavonoid Test: The extract (0.2 g) was gently warmed in ethanol and filtered. Two pieces of magnesium chips were then added, followed by 2 mL of concentrated HCl. No observable color change indicated the absence of flavonoids.

Sodium Hydroxide Test: The extract (0.2 g) was dissolved in 2 mL of distilled water and filtered. Then, 1 mL of 5% sodium hydroxide solution was slowly added. For both extracts, the solution changed from dark

green to yellow, indicating the presence of flavonoids.

Saponin Test: The extract (0.2 g) was placed in a test tube with 5 mL of distilled water, shaken vigorously for 60 seconds, and left to stand for 15 minutes. The presence of persistent frothing on the surface confirmed the presence of saponins.

Alkaloid Test: The extract (0.2 g) was dissolved in ethanol, followed by the addition of 5 mL of 1% diluted HCl. The mixture was warmed and filtered. Then, 2 mL of the filtrate was placed in a test tube, and Dragendorff's reagent was added. The appearance of a reddish color indicated the presence of alkaloids.

Tannin Test: The extract (0.2 g) was mixed with 5 mL of distilled water, stirred, and filtered. To 2 mL of the filtrate, two drops of 1% ferric chloride solution were added and shaken. The formation of a black-colored solution confirmed the presence of tannins.

Cardiac Glycoside Test: The extract (0.2 g) was dissolved in chloroform, and 1 mL of concentrated sulfuric acid (H₂SO₄) was carefully added down the side of the test tube. The presence of a reddish-brown color at the interface indicated the presence of a steroidal ring, characteristic of the aglycone portion of cardiac glycosides.

Anthraquinone Test: The extract (0.2 g) was mixed with distilled water, shaken with 3 mL of benzene, and filtered. To the filtrate, 5 mL of 10% ammonia solution was added, and the mixture was shaken for 60 seconds. No significant color change was observed, indicating the absence of anthraquinones.

Quantitative Phytochemical Screening

Total Phenolic Content

The total phenolic content (TPC) was determined spectrophotometrically using a modified version of the method by Kim *et al.* (2003) (16). A 0.5 mL sample (1 mg/mL in methanol) was mixed with 2.5 mL of 10% Folin-Ciocalteu reagent and 2 mL of 7% Na₂CO₃. The mixture was vortexed for 15 seconds and incubated in the dark at 40 °C for 30 minutes to allow color development. Absorbance was measured at 765 nm using a UV-Vis spectrophotometer (Model: Techmel and Techmel, USA). A calibration curve was generated using gallic acid solutions (10–100 µg/mL), and TPC was calculated by extrapolating sample absorbance values from the standard curve. Results were expressed as mg of gallic acid equivalent per gram of dry weight.

Total Flavonoids Content

The total flavonoid composition was obtained utilizing the protocol of Subhashini *et al.* (2010). A 1 mg/mL extract solution was diluted with 200 µL of distilled water, followed by the addition of 150 µL of 5% sodium

nitrite (NaNO₂) solution. This mixture was incubated for 5 minutes and added to 150 µL of 10% AlCl₃·6H₂O. After 6 minutes, 2 mL of 1M NaOH was added. The absorbance reading was noted using a UV-Vis spectrophotometer at 510 nm and the total flavonoid content was expressed as mg of quercetin (QE) equivalent per gram dry weight (17).

Radical Scavenging Activity Assay

The antioxidant activity of the extract was assessed using the DPPH radical scavenging assay, following a standard methodology (18). A 1 mL aliquot of 0.1 mM DPPH solution was mixed with 3 mL of the test solution containing the extract and ascorbic acid. The mixture was stirred for one minute and then incubated in the dark for 30 minutes. Absorbance was measured at 517 nm using a UV-Vis spectrophotometer (Techmel and Techmel, USA).

Ferric Reducing Antioxidant Power Assay

The ferric reducing power was determined according to the method of Ali *et al.* (2020). Various concentrations (µg/mL) of the extract were added to (1 mL) of 200 mM sodium phosphate buffer (pH 6.6) and 1 mL of (0.69 mL) potassium ferricyanide, [K₃[Fe(CN)₆]. The mixture was incubated at 50 °C for 20 minutes. Thereafter, trichloroacetic acid (1 mL, 10%) was prepared and dissolved with 50 mL of distilled water. The mixture was centrifuged at 650 rpm for 10 minutes. The upper layer (4 mL) was mixed with (4 mL) of deionised water and 0.8 mL of 0.1% of ferric chloride anhydrous (FeCl₃), and the absorbance was measured with UV-Vis spectrophotometer at 700 nm. The same procedure was repeated with various concentrations of ascorbic acid. Higher absorbance indicates higher reducing power (19).

Experimental Animals

Albino Wistar rats (120 -135 g) of either sex maintained at the animal house of the Department of Pharmacology and Toxicology, Faculty of Pharmacy, University of Uyo, Nigeria, were used for the study. The animals were housed in standard cages and were maintained on standard pelleted feed (Guinea feed) and water *ad libitum*. The approval for animal studies was obtained from the College of Health Sciences Animal Ethics Committee, University of Uyo (UU/CS/AE/14/63).

Alpha-Amylase Inhibitory Study Using Starch as Substrate

Thirty Wistar rats were randomly divided into six groups, with five rats per group. All rats were fasted for 18 hours before the experiment, and their fasting blood glucose levels were measured at 0 minutes (before administration).

- **Group I (Normal Control):** Received distilled

water (10 mL/kg).

- **Group II:** Administered starch (2 g/kg) with distilled water (10 mL/kg) as a vehicle.
- **Group III:** Administered starch (2 g/kg) along with the standard drug acarbose (100 mg/kg).
- **Groups IV, V, and VI:** Administered starch (2 g/kg) along with *M. barteri* leaf extract at doses of 150, 300, and 450 mg/kg, respectively.

All substances were administered orally. Blood glucose levels were monitored at 60, 90, 120, and 180 minutes to assess the effect of the extract on enzyme activity.

Sucrose Inhibitory Study (Using Sucrose as Substrate)

Thirty Wistar rats were randomly divided into six groups, with five rats in each group. All rats were fasted for 18 hours before the experiment, and their fasting blood glucose levels were measured at 0 minutes (before administration).

- **Group I (Normal Control):** Received distilled water (10 mL/kg).
- **Group II:** Administered sucrose (2 g/kg) with distilled water (10 mL/kg) as a vehicle.
- **Group III:** Administered sucrose (2 g/kg) along with the standard drug acarbose (100 mg/kg).
- **Groups IV, V, and VI:** Administered sucrose (2 g/kg) along with *M. barteri* leaf extract at doses of 150, 300, and 450 mg/kg, respectively.

All substances were administered orally. Blood glucose levels were monitored at 60, 90, 120, and 180 minutes to evaluate the effect of the extract on enzyme activity.

Maltose Inhibitory Study (Using Maltose as Substrate)

Thirty Wistar rats were randomly divided into six groups, with five rats in each group. All rats were fasted for 18 hours before the experiment, and their fasting blood glucose levels were measured at 0 minutes (before administration).

- **Group I (Normal Control):** Received distilled water (10 mL/kg).
- **Group II:** Administered maltose (2 g/kg) with distilled water (10 mL/kg) as a vehicle.
- **Group III:** Administered maltose (2 g/kg) along with the standard drug acarbose (100 mg/kg).
- **Groups IV, V, and VI:** Administered maltose (2 g/kg) along with *M. barteri* leaf extract at doses of 150, 300, and 450 mg/kg.

All administrations were performed orally. Blood glucose levels were monitored at 60, 90, 120, and 180

minutes to evaluate the effect of the extract on enzyme activity.

Blood Glucose Determination

Blood glucose concentration was measured using a glucometer (Fine Test) according to the manufacturer's specifications. Blood samples were obtained by pricking the tip of each rat's tail, and a drop of blood was applied to a test strip.

The glucometer operates using an electrochemical detection system based on the glucose oxidase method. Each disposable reagent strip contains an electrode impregnated with glucose oxidase, which reacts with glucose in the blood sample upon contact with the membrane covering the reagent pad. This reaction produces gluconic acid and generates an electric current. An electrochemical mediator transfers electrons to the electrode surface, and the sensor measures the resulting current. The magnitude of the generated current is proportional to the glucose concentration in the blood sample, providing an accurate measurement of blood glucose levels (21).

Statistical Analysis

Data obtained from this work were analysed statistically using one-way ANOVA followed by Tukey-Kramer multiple comparison test using InStat Graphpad software, (San Diego, USA). Differences between means were considered significant at $p < 0.05$ and very significant at $p < 0.001$.

Results and Discussion

Plant Extraction

Extraction of *M. barteri* leaves with methanol afforded methanol crude extract (9.5% yield). However, aqueous extraction of the leaves gave 14% yield of the aqueous extract. The results showed that aqueous extraction afforded the highest yield showcasing the effect of solvent polarity (see **Table 1**).

Table 1. Yield of methanol and aqueous extracts.

Extract	Sample (g)	Extract (g)	Yield (%)
Methanol	90	8.6	9.5
Aqueous	90	12.3	14

Preliminary Phytochemical Screening

The results of the preliminary phytochemical screening of *M. barteri* leaf extracts showed the presence of alkaloid, flavonoid, saponins, tannins and cardiac glycosides in the methanol and aqueous extracts while the presence of anthraquinones were not detected in both extracts during the preliminary phytochemical analysis (see **Table 2**).

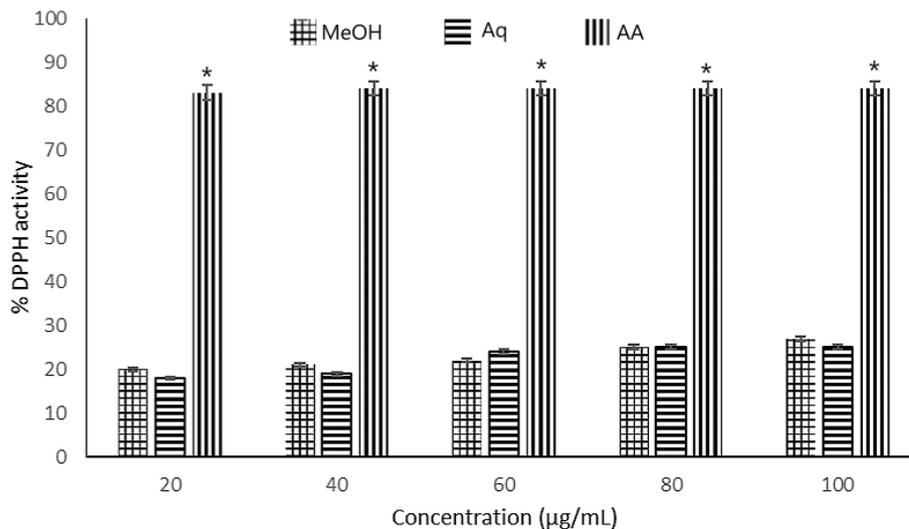


Figure 1. DPPH radical scavenging activity of different concentrations (20-100 µg/mL) of *Maesobotrya barteri* leaves methanol and aqueous extracts. **Note:** MeOH = methanol extract; Aq = aqueous extract; AA = ascorbic acid; and DPPH = 1,1-diphenyl-2-picryl-hydrazyl free radical. (*, $p < 0.05$) indicates a significant difference compared to other groups.

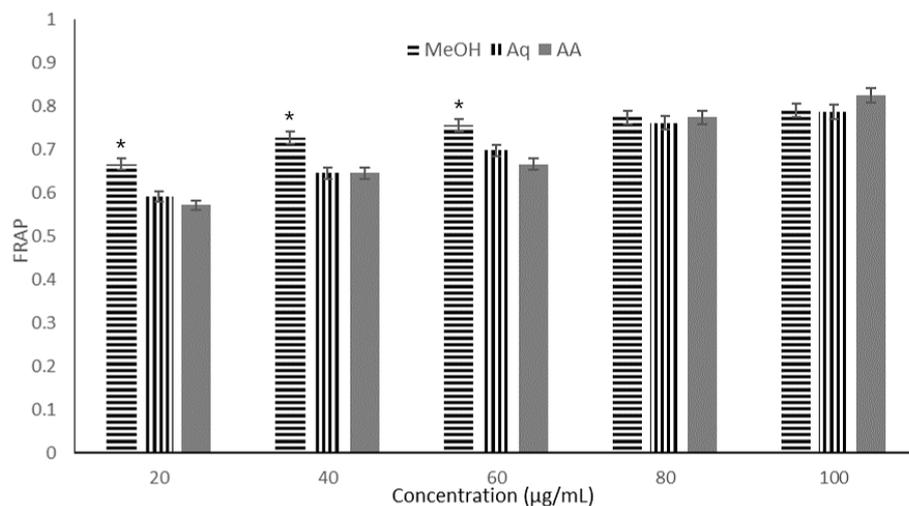


Figure 2. Ferric reducing antioxidant power (FRAP) activity of different concentrations (20-100 µg/mL) of *Maesobotrya barteri* leaves methanol and aqueous extracts. **Note:** MeOH = methanol extract; Aq = aqueous extract; AA = ascorbic acid; and DPPH = 1,1-diphenyl-2-picryl-hydrazyl free radical. (*, $p < 0.05$) indicates a significant difference compared to other groups.

Table 2. Preliminary phytochemical screening of methanol and aqueous extracts.

Phytochemicals	Results	
	Methanol	Aqueous
Alkaloids	+	+
Flavonoids	+	+
Saponins	+	+
Tannins	+	+
Anthraquinones	-	-
Cardiac Glycosides	+	+

Note : Key: + = Present; - = absent.

Total Phenolic and Flavonoid Contents

The total phenolic content in the extracts ranged from 0.04 mg GAE/g to 3.18 mg GAE/g. From the gallic acid calibration curve expression ($y = 0.014x + 0.1625$), the aqueous extract of *M. barteri* showed the highest phenolic content (3.18 mg GAE/g) compared to the methanol extract (0.04 mg GAE/g). The reverse was observed for the total flavonoid content of the extracts. The aqueous extract had the least flavonoid content (27.70 mg QE/g) while the methanolic extract revealed the highest flavonoid content (57.70 mg QE/g), representing about a 2-fold increase in the total flavonoid content (see **Table 3**).

Table 3. Total phenolics, flavonoids, DPPH values, and FRAP values of methanol and aqueous extracts of *Maesobotrya barteri* leaves.

Assay	Methanol	Aqueous	Ascorbic acid
TPC (mg GAE/g)	0.04	3.18	
TFC (mg QE/g)	57.70	27.70	
DPPH IC ₅₀ (µg/mL)	192.95	196.04	17.19
FRAP IC ₅₀ (µg/mL)	30.48	37.64	38.15

Radical Scavenging and Ferric Reducing Antioxidant Power Assay

Both extracts scavenged radical of DPPH and exhibited reducing potential in a concentration-dependent manner (see **Figure 1** and **2**). The methanolic extract scavenged radicals of DPPH with higher inhibition percentage compared to the aqueous extract even though the effect was not so remarkable. At 100 µg/mL concentration, methanol extract scavenged 27% of DPPH radicals while the aqueous extract was able to scavenge 25% of DPPH radicals. The standard drug (ascorbic acid) exhibited 84% inhibitory activity. At 40 µg/mL, the aqueous extract scavenged 21% of the DPPH radicals comparable to the 19% scavenging activities of the methanol extract. For DPPH, the IC₅₀ values of the methanol and aqueous extracts were 192.50 µg/mL and 196.04 µg/mL respectively whereas ascorbic acid exhibited IC₅₀ value of 17.19 µg/mL. For FRAP, again methanol extract exhibited the highest reducing power (IC₅₀ = 30.48 µg/mL, aqueous extract demonstrated the least FRAP activity (IC₅₀ = 37.64 µg/mL). Ascorbic acid, the standard drug showed IC₅₀ value of 38.15 µg/mL, indicating that methanol extract exhibited the highest reducing power. The antioxidant activity of extracts also showed strong correlation with the total flavonoids and phenolics in the extracts (see **Table 4**).

Table 4. Correlation between antioxidant assays, total phenolics and total flavonoid.

	DPPH	FRAP	TP	TF
DPPH	1			
FRAP	0.99	1		
TP	0.79	0.96	1	
TF	0.79	0.95	0.99	1

Administration of starch (2 g/kg) to fasted rats caused varying percentages of increase in blood glucose levels of the treated animals after one hour. The percentages were starch (40.31%), *M. barteri* leaf extract-treated groups (11.73 - 16.78%), and acarbose-treated group (11.06%). These increases were reduced after 120 minutes with the group treated with the low and middle doses of the extract (500 and 1000 mg/kg) having percentage increases of 1.44 and 7.68 % respectively. All the extract-treated groups had their BGL reduced to a normal level at 180 minutes and this was sustained throughout the study. Also, co-administration of the starch with acarbose prominently inhibited the rise in the blood glucose concentrations (see **Table 5**).

Administration of maltose (2 g/kg) to fasted rats caused varying percentages of increase in blood glucose levels of the treated animals after one hour. The percentages were maltose (44.95%), *M. barteri* leaf extract-treated groups (1.29 - 41.39%), and acarbose-treated group (0.77%). These increases were reduced to normal after 120 minutes and this was sustained throughout the duration of the study. Also, co-administration of the maltose with acarbose prominently inhibited the rise in the blood glucose concentrations (see **Table 6**).

Table 5. Effect of methanol leaf extract of *Maesobotrya barteri* on blood glucose level of rat after oral administration of starch load.

Treatment	Dose (mg/kg)	Blood glucose level (mg/dL, [%]) in minutes			
		0 min	60 min	120 min	180 min
Control (normal saline)	-	86.00 ± 11.53	87.66 ± 7.62 [1.93]	91.0 ± 7.50 [5.81]	80.00 ± 6.02
Starch (negative control)	2000	80.0 ± 4.54	112.25 ± 4.73 [40.31]	92.50 ± 1.70 [15.62]	87.25 ± 6.52 [9.06]
Acarbose (positive control)	100	72.33 ± 2.69	80.33 ± 7.21 [11.06]	74.0 ± 1.00 [2.30]	72.33 ± 8.68
	150	95.33 ± 4.48	111.33 ± 11.31 [16.78]	102.66 ± 2.33 ^a [7.68]	90.0 ± 4.00 ^a
Extract	300	92.66 ± 0.33	104.0 ± 11.54 [11.73]	94.0 ± 6.24 ^b [1.44]	80.0 ± 3.00 ^a
	450	89.33 ± 2.02	102.66 ± 8.09 [14.92]	105.0 ± 6.65 ^a [17.54]	88.66 ± 6.76

Note: Data is expressed as MEAN ± SEM, significant at ^a*p* < 0.05 and ^b*p* < 0.01, when compared to control (n=6). Values in brackets are percentage increases in blood glucose concentrations compared to 0 min in the same group.

Table 6. Effect of methanol leaf extract of *Maesobotyra barteri* on blood glucose level of rat after oral administration of maltose load.

Treatment	Dose (mg/kg)	Blood glucose level (mg/dL, [%]) in minutes			
		0 min	60 min	120 min	180 min
Normal control	-	86.00 ± 11.53	87.66 ± 7.62 [1.93]	91.0 ± 7.50 [5.81]	80.00 ± 6.02
Maltose (negative control)	-	86.75 ± 2.52	125.75 ± 1.65 [44.95]	99.50 ± 2.90 [12.75]	88.0 ± 1.68 [1.44]
Acarbose (positive control)	100	85.34 ± 1.36	86.0 ± 2.20 [0.77]	84.26 ± 1.14 ^a	82.28 ± 2.26
Extract	150	102.33 ± 4.70	103.66 ± 3.28 [1.29]	87.33 ± 1.66	78.0 ± 4.93
	300	94.66 ± 5.36	122.33 ± 7.53 [29.23]	82.66 ± 2.60	79.33 ± 3.18
	450	100.66 ± 7.53	142.33 ± 7.31 [41.39]	92.0 ± 7.23	88.0 ± 14.84

Note: Data is expressed as MEAN ± SEM, significant at ^a $p < 0.05$, ^b $p < 0.01$, when compared to control (n=6). Values in brackets are percentage increases in blood glucose concentrations compared to 0 min in the same group.

Table 7. Effect of methanol leaf extract of *Maesobotyra barteri* on blood glucose level of rat after oral administration of sucrose load.

Treatment	Dose (mg/kg)	Blood glucose level (mg/dL, [%]) in minutes			
		0 min	60 min	120 min	180 min
Normal control	-	86.00 ± 11.53	87.66 ± 7.62 [1.93]	91.0 ± 7.50 [5.81]	80.00 ± 6.02
Sucrose (negative control)	-	81.0 ± 4.50	112.66 ± 1.49 ^a [39.08]	97.33 ± 1.63 [20.16]	94.15 ± 4.81 [16.23]
Acarbose (positive control)	100	90.33 ± 2.48	82.0 ± 6.00	71.66 ± 3.75	78.0 ± 3.78
Extract	150	93.33 ± 2.18	92.33 ± 3.52	85.33 ± 3.18	76.0 ± 2.08
	300	84.66 ± 1.84	97.33 ± 0.33 [14.96]	89.33 ± 1.20 [5.51]	67.66 ± 3.71
	450	90.66 ± 7.83	101.66 ± 5.17 [12.13]	85.33 ± 4.17	81.66 ± 3.81

Note: Data is expressed as MEAN ± SEM, significant at ^a $p < 0.05$, ^b $p < 0.01$, when compared to control (n = 6). Values in bracket are percentage increases in blood glucose concentrations compared to 0 min in the same group.

Administration of sucrose (2 g/kg) to fasted rats caused varying percentages of increase in blood glucose levels of the treated animals after one hour. The percentages were sucrose (39.08%), and *M. barteri* leaf extract-treated groups (0 - 14.96%). The low dose (500 mg/kg) and acarbose-treated groups had no increment in BGL. These increases were reduced to normal after 120 minutes with only the group treated with the middle dose (1000 mg/kg) of the leaf extract having BGL increment of 5.51 %. All the groups treated with the leaf extract had their BGL reduced to a normal level at 180 minutes and this was sustained throughout the study (see **Table 7**).

Plants are recognized for producing a variety of chemical compounds which have numerous commercial and industrial uses, including flavours, aromas, fragrances, enzymes, preservatives,

cosmetics, natural pigments, and bioactive compounds. Research into these phytochemicals has led to the isolation and identification of many natural products, such as phenolics, flavonoids, organic acids, steroids, terpenes, lignans, glycosides, and alkaloids (2). Among these, flavonoids, phenolic acids, and tocopherols are particularly valued for their potent antioxidant properties, which help scavenge free radicals in the human body (22). This study examined the *in vitro* antioxidant activity of methanol and aqueous extracts from *M. barteri* leaves, as well as their effects on blood glucose levels. We assessed the extracts' ability to scavenge the 2,2-Diphenyl-1-picrylhydrazyl (DPPH) radical, their ferric reducing antioxidant power, total phenolic and flavonoid content, and their inhibitory effects on the alpha-amylase and alpha-glucosidase in Wistar rats.

In terms of total phenolic content, the aqueous extract had higher content (7.68 mg GAE/g) compared to the methanol extract, while total flavonoid content was higher in the methanolic extract (271.40 mg QE/g), suggesting that extraction with methanol afforded a more potent path to flavonoids. In the DPPH scavenging assay, both extracts demonstrated radical scavenging activity against DPPH radicals, the methanolic extract showed a slightly higher inhibition percentage than the aqueous extract although the difference was minimal. At 60 $\mu\text{g/mL}$, the aqueous extract scavenged 24% of DPPH radicals, while the methanolic extract scavenged 22%. At 100 $\mu\text{g/mL}$, the methanol extract scavenged 27% of DPPH radicals, compared to 25% by the aqueous extract. The standard drug, ascorbic acid, demonstrated a much higher inhibition (84%), as shown in Figure 2. The IC_{50} values, derived from plotting concentration ($\mu\text{g/mL}$) against percentage inhibition for DPPH, were: methanol extract (192.95 $\mu\text{g/mL}$), aqueous extract (196.04 $\mu\text{g/mL}$), and ascorbic acid (17.19 $\mu\text{g/mL}$), as listed in Table 2. These results indicate that both the methanolic and aqueous extracts of *Maesobotrya* leaves showed weak DPPH radical scavenging activity compared to the control (ascorbic acid), although the methanol extract exhibited a better scavenging radical activity. The high IC_{50} values ($>100 \mu\text{g/mL}$) suggest that these extracts do not possess strong antioxidant activity. The ferric reducing antioxidant power was evaluated by plotting mean absorbance values against extract concentrations ($\mu\text{g/mL}$) in the range of 20-100 $\mu\text{g/mL}$ (see **Figure 3**). At 40 $\mu\text{g/mL}$, the absorbance values were 0.727, 0.645, and 0.571 for methanol, aqueous, and ascorbic acid, respectively. At 100 $\mu\text{g/mL}$, the absorbance values increased to 0.790, 0.786, and 0.824 for methanol, aqueous, and ascorbic acid, respectively. The IC_{50} values were: methanol extract (30.48 $\mu\text{g/mL}$), aqueous extract (37.64 $\mu\text{g/mL}$), and ascorbic acid (38.15 $\mu\text{g/mL}$), as shown in Figure 2. In terms of the reducing power the results demonstrate that the methanol extract had a higher reducing power than the aqueous extract. Ferric reducing assay has been described as one of the useful tools in profiling antioxidant capabilities (4, 34). The relatively low phenolic and flavonoid content in the methanol extract (0.04 mg GA/g and 57.70 mg QE/g) and the aqueous extract (3.18 mg GA/g and 27.70 mg QE/g) may explain the observed weak antioxidant activity, suggesting a correlation between the plant's antioxidant properties and its polyphenolic content. Total phenolics showed good correlation with DPPH activity ($r^2 = 0.79$) and reducing power ($r^2 = 0.96$) respectively. Similar correlation was observed for the flavonoids demonstrating the correlation between antioxidant properties and polyphenolic compounds (see **Table 4**).

The methanol extract significantly inhibited

increases in blood glucose concentrations following starch administration, independent of the dose. Starch is a vital carbohydrate source for humans and other animals, and its digestion primarily involves two enzymes: α -amylase and α -glucosidase. α -amylase, found in saliva and pancreatic juices, breaks down starch by cleaving α -(1 \rightarrow 4)-D-glycosidic bonds, producing smaller oligosaccharides and disaccharides (23). These disaccharides are further reduced to monosaccharides by membrane-bound α -glucosidase enzymes (24). Inhibition of these enzymes slows carbohydrate digestion, resulting in a more gradual increase in blood glucose levels, as observed in our study.

Research indicates that reducing postprandial hyperglycemia can effectively manage early-stage of diabetes by inhibiting α -glucosidase and α -amylase, which slows glucose absorption and mitigates plasma glucose spikes after meals (25-27). Acarbose, voglibose, and miglitol are commercial antidiabetic medications however, these medications cause side effects such as flatulence, stomach discomfort, and allergic reactions (28). The inhibitory effects of plant extracts on alpha amylase and alpha glucosidase enzymes activities have been documented (29, 30) and has been linked to the bioactive constituents in the extracts (31, 32). Many works have shown that various plant bioactive compounds possess hypoglycemic properties, affecting multiple targets, proteins, and enzymes (33-35). Alkaloids, flavonoids, anthocyanins, terpenoids, phenolic compounds, and glycosides have been identified as bioactive α -glucosidase inhibitors with significant impacts on diabetes management. These compounds influence glucose transportation in the body and intestinal glycosidase activity (36-38). Additionally, polyphenols act as metal ion chelators and protein precipitation agents (37, 38).

Notably, when co-administered with maltose and sucrose, the methanol extract of *M. barteri* significantly inhibited blood glucose increases, albeit lower compared to acarbose. Phytochemical screening revealed that both extracts contained flavonoids, alkaloids, saponins, tannins, and glycosides. Tannins, flavonoids, terpenoids, and phenolic acids are known for their strong antioxidant, antidiabetic, and anti-inflammatory properties (8, 39). Flavonoids exist in various forms, including flavones, isoflavones, flavanones, and flavanols and are reported to inhibit α -glucosidase. For instance, as shown in **Figure 3**, Calodenin A and (-)-epigallocatechin-gallate (compound **1** & **2**) exhibit significant α -glucosidase inhibition (IC_{50} : 0.4 μM and IC_{50} : 5.2 μM , respectively) compared to the control, acarbose (IC_{50} : 93.6 μM and IC_{50} : 1400 μM) respectively, (40, 41). Similarly, procyanidin A2 (compound **3**) demonstrated efficacy in managing postprandial diabetes with an IC_{50} of 0.27

g/mL (42). Phenolic acids such as caffeic acid and vanillic acid (compound **4** & **5**) also showed superior α -glucosidase inhibition compared to standard drugs, acarbose (IC_{50} : 0.39 and IC_{50} : 8.38, respectively) (42). Furthermore, a study by Chen et al. (2020) reported that a diterpenoid, taxumariene F (compound **6**) showed better inhibitory activity, with an IC_{50} of $3.7 \pm 0.75 \mu\text{M}$ when compared to the standard antidiabetic drug, acarbose (IC_{50} of $155.86 \pm 4.12 \mu\text{M}$). Their findings also indicated that the activity of the diterpenoid was due to the presence of the tricycle,

epoxy and acetoxy groups in the compound (43). While the inhibition of alpha amylase and alpha glucosidase study helps to understand the mechanism of antidiabetic activity, the antioxidant potential of a plant extract could enhance its antidiabetic effects by lowering oxidative stress, boosting insulin sensitivity, reducing inflammation, and safeguarding beta-cells from destruction by free radicals generated in the body in diabetic condition (31-36; 44). These all play a crucial role in regulating blood sugar levels and preventing complications of diabetes.

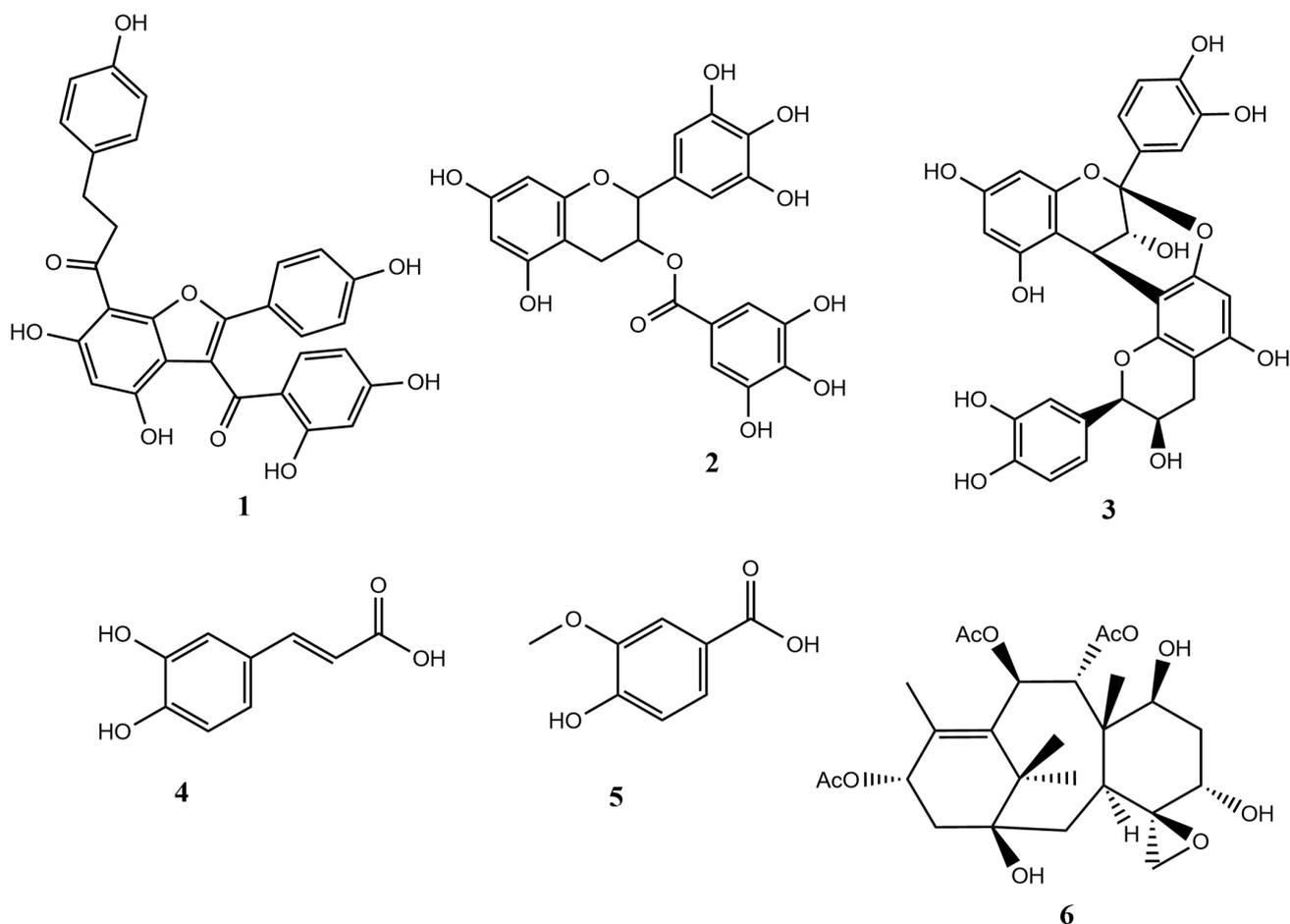


Figure 3. 2D chemical structure of alodenin A (compound **1**), (-)- epigallocatechin-gallate (compound **2**), procyanidin A2 (compound **3**), caffeic acid (compound **4**), vanillic acid (compound **5**), and taxumariene (compound **6**).

Flavonoids and other polyphenols are recognized for their antioxidant properties and role in managing postprandial hyperglycemia, while tannins significantly influence carbohydrate metabolism. Terpenoids contribute to plant growth, development, and stress resilience (28, 45). Given the increasing prevalence of diabetes, investigating natural products is a promising avenue for developing new hypoglycemic treatments. Future research should aim to isolate and characterize these bioactive compounds, examining the understanding of their molecular mechanisms to facilitate the design of more effective antidiabetic derivatives. Additionally, exploring the synergistic

effects of these compounds in conjunction with existing antidiabetic therapies may lead to innovative strategies for managing postprandial diabetes.

Conclusion

The findings of this study suggested that the leaves extracts of *M. barteri* possess phytochemicals such as phenols, flavonoids and tannins which are known free radical scavengers. Their ability to scavenge free radicals in the antioxidant analysis could be implicated in the antidiabetic study in which the methanol extract showed inhibition of α -amylase and α -glucosidase

enzymes being one of the modes of antidiabetic activity.

Declarations

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Conflict of Interest

The authors declare no conflicting interest.

Data Availability

The unpublished data is available upon request to the corresponding author.

Ethics Statement

The approval for animal studies was obtained from the College of Health Sciences Animal Ethics Committee, University of Uyo (UU/CS/AE/14/63). Ethical conditions guidelines for the care and use of Laboratory Animals, governing the conduct of experiments with life animals, were strictly maintained.

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